

BCF-003-1015005

Seat No. _____

B. Sc. (Sem. V) (CBCS) (W.E.F. 2016) Examination

August - 2021

C - 501 : Chemistry

(Inorganic Chemistry and industrial Chemistry)
(New Course)

Faculty Code: 003

Subject Code: 1015005

Time : $2\frac{1}{2}$ Hours] [Total Marks: 70 Instructions: (1) There are ten questions. Answer any five questions. (2) All questions carry equal marks (14 marks) (3)Figure to the right indicates full marks. 1 (A) Answer the following questions: 4 What is zero point energy? Give the equation of Hamiltonian operator for a particle in one dimensional system. (3) Give equation of orthogonality condition. **(4)** Define operator. (B) Explain multiplication of operator. 2 3 (C) Explain Hamiltonian operator. (D) Discuss in detail particle moving in one dimensional box. 5 2 (A) Answer the following questions: 4 Give the operator of addition (1) (2) $\hat{A} = \log_e \& \hat{B} = \frac{d}{dx}, f(x) = x^2 \text{ then } (\widehat{A+B}) f(x) =$

(4) Write down ∇ and ∇^2 .

	(B)	Define substraction of operators.	2
	(C)	What is degeneracy? Calculate degeneracy of energy	3
		level, where $E = \frac{144^2}{8ma^2}$ in cubic box of length a.	
	(D)	Give the schrodinger equation in polar coordinates	5
		and derive R , θ , ϕ variable separation.	
3	(A)	Answer the following questions:	4
		(1) From $[Ni(CN)_4]^{-2}$ and $[Ni(Cl)_4]^{2-}$ which one is diamagenetic.	
		(2) Who explained the magnetic properties of complexes using CFT?	
		(3) Give the equation of CFSE for tetrahedral field in terms of Δ_t and Δ_o both.	
		(4) Give the equation for calculate μ_s .	
	(B)	Give the factors affecting splitting energy.	2
	(C)	Write short note on splitting of d-orbital in tetrahedral field.	3
	(D)	Discuss about the contribution of orbital angular	5
		momentum towards the magnetic momentum of 3d-metal complex.	
4	(A)	Answer the following questions :	4
		(1) For $n = 2$; what is μ_s ?	
		(2) Calculate CFSE for d^1 system in octahedral ligand field.	
		(3) Give the full form of CFSE.	
		(4) An atom or group of atoms that donate its lone pair	
		of electrons into metal orbitals is known as	
	(B)	Give the splitting of d-orbitals in presence of octahedral and tetrahedral ligand field by figure only.	2
	(C)	Explain the strength of ligands.	3
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(D) Discuss splitting of d-orbitals in octahedral ligand field.

5

5	(A)	Answer the following questions:	4
		(1) Define Ω – acid ligands.	
		(2) $Ni(CO)_4$ is in shape by hybrids.	
		(3) Give the full form of RCC.	
		(4) What is slag cement?	
	(B)	Explain the classification of metal carbonyls.	2
	(C)	Give the difference between dry and wet process of cement.	3
	(D)	Write short note on structure of Fe (CO) $_5$.	5
6	(A)	Answer the following questions:	4
		(1) Define metal nitrosyls.	
		(2) Which type of hybridization in MN_2 (CO) ₁₀ ?	
		(3) Define water proof cement.	
		(4) Give the name and formula of C_3S .	
	(B)	Give the uses of cement.	2
	(C)	Give the reaction for preparation of $\operatorname{Fe}_3(\operatorname{CO})_{12}$ and	3
		draw the molecular structure.	
	(D)	Discuss about types of cement.	5
7	(A)	Answer the following questions:	4
		(1) Define fertilizers.	
		(2) Give the full form of DAP.	
		(3) What is complete and incomplete fertilizer	
		(4) Write secondary nutrients of fertilizers.	
	(B)	What is the effect of deficiency of N and P on plant growth?	? 2
	(C)	Give the process flow diagram of manufacture of	3
		ammonium sulphate from gypsum.	
	(D)	What is phosphate fertilizer? List their names and	5
		describe manufacturing of any one of them.	
8	(A)	Answer the following questions:	4
		(1) What is micro nutrient?	
		(2) Write primary nutrients of fertilizers?	
		(3) Write the formula (structure) of triple super phosphate.	
		(4) Give the name of natural inorganic fertilizers.	
	(B)	Give the classification of fertilizers according to its sources.	2
BCI	7-003 -	[Contd.	••••

	(C)	List potassium fertilizer and give chemical reaction	3
		involved in manufacturing of KNO_3 .	
	(D)	List nitrogenous fertilizers and discuss production	5
		method of ammonium sulphate by sindri process with its	
		action as fertilizer.	
9	(A)	Answer the following questions :	4
		(1) Define glass.	
		(2) Which compounds are used for decolourise glass?	
		(3) Give the chemical formula of cryolite.	
		(4) What is cullet?	
	(B)	Give the physical properties of glass.	2
	(C)	Describe photochemical glass.	3
	(D)	Discuss about the manufacturing process of glass.	5
10	(A)	Answer the following questions:	4
		(1) What is annealing?	
		(2) Give the name of chief ingredient of glass.	
		(3) Write two oxidizing agents used in glass	
		manufacturing.	
		(4) Write the chemical structure of Dolomite.	
	(B)	Give the properties of borosilicate glass.	2
	(C)	Give the chemical reactions involved in glass	3
		manufacturing.	
	(D)	Define glass. List the raw materials for glass	5
		with their functioning and importance.	